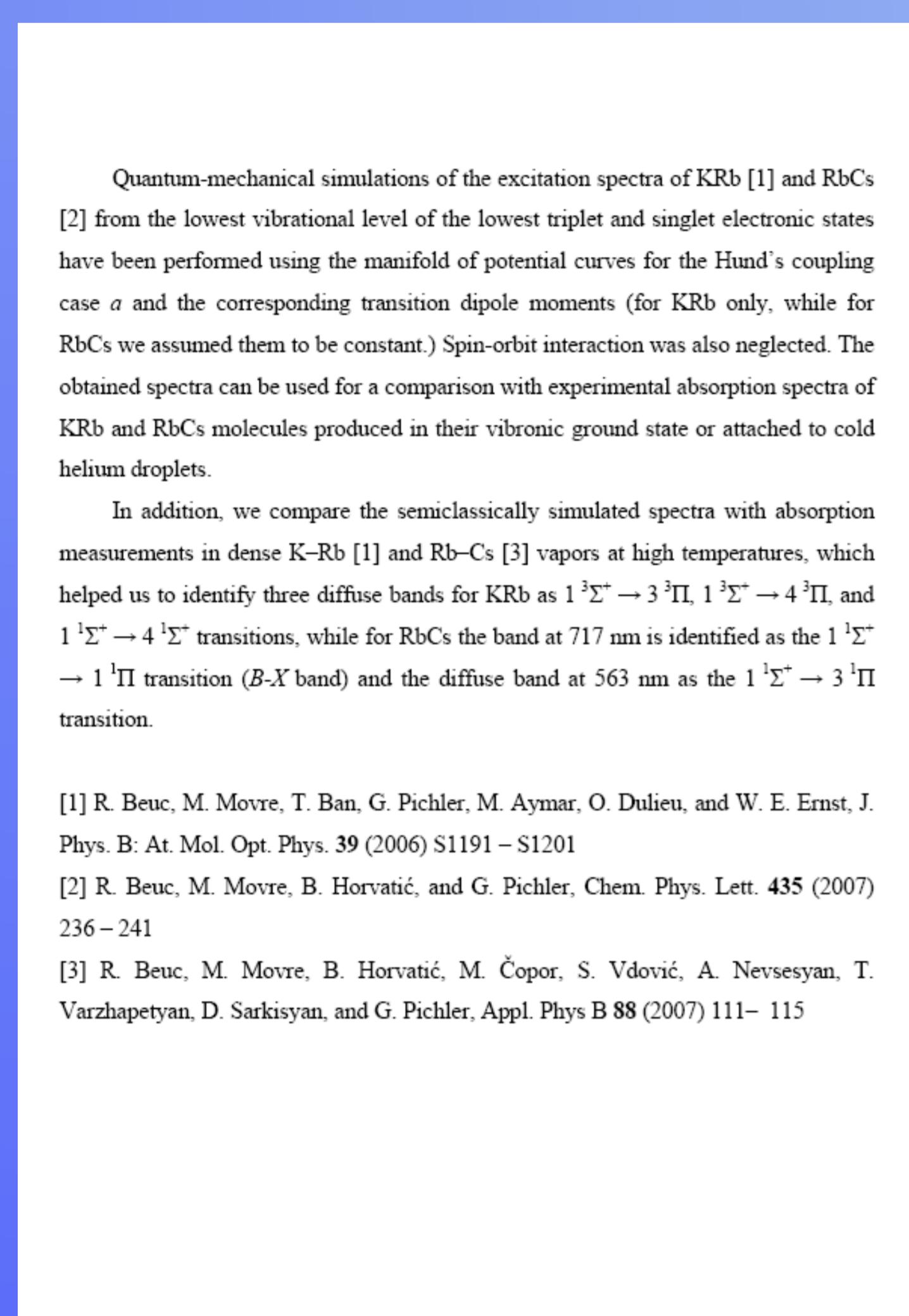


Predictions of KRb and RbCs spectra under cold and thermal conditions

Robert Beuc, Berislav Horvatić, and Mladen Movre

Institute of Physics, Bijenička cesta 46, P.O.Box 304, HR-10001 Zagreb, Croatia

beuc@ifs.hr horvatic@ifs.hr movre@ifs.hr



Theoretical simulations

The absorption coefficient from a single rovibrational level (v'', J'', Λ'') of the lower electron state Λ'' into all upper rovibrational levels (v', J', Λ') of the excited electron state Λ' :

$$k(v) = N \sum_{v' J' \Lambda'} \sigma_{v' J' \Lambda'}^{v'' J'' \Lambda''} (v), \quad (1)$$

where N is the concentration of molecules in the initial level and $\sigma_{v' J' \Lambda'}^{v'' J'' \Lambda''}(v)$ is the absorption cross section for the transition from (v'', J'', Λ'') to (v', J', Λ'), given by

$$\sigma_{v' J' \Lambda'}^{v'' J'' \Lambda''}(v) = \frac{8\pi k}{3hc} \left| \langle \phi_{v' J' \Lambda'} | D(R) | \phi_{v'' J'' \Lambda''} \rangle \right|^2 g(v - v_g) \frac{s_{v' J' \Lambda'}^{v'' J'' \Lambda''}}{2J'+1}, \quad (2)$$

where $D(R)$ is the dipole transition moment, $s_{v' J' \Lambda'}^{v'' J'' \Lambda''}$ is the Hönl-London factor, v_g is the transition frequency ($v_g = E_{v' J' \Lambda'} - E_{v'' J'' \Lambda''}$), and $g(v - v_g)$ is the line-shape function of the molecular transition.

At the temperature of helium nanodroplets ($T = 0.4$ K) only a few rotational levels within the lowest vibrational level ($v'' = 0$) of a free RbCs molecule are populated. \Rightarrow Neglect the J dependence of the matrix elements and put $J'' = J' = 0$

$$k(v) = \frac{8\pi k}{3hc} N \sum_v |\langle \phi_0 | D(R) | \phi_v \rangle|^2 g(v - v_g), \quad (3)$$

$\phi_{v' J' \Lambda'} \equiv \phi_0$ is the lowest rovibrational wave function ($v'' = 0$, $J'' = 0$, $\Lambda'' = 0$).

\Rightarrow Energies and wave functions for vibrational levels determined by using the Fourier Grid Hamiltonian (FGH) method

To get a coarse-grained spectrum \Rightarrow replace the summation over v' in (3) with an integral over (quasicontinuous) energy E \Rightarrow Turn the unity-normalized bound state wave function into an energy-normalized one, $\phi_E = |E/\partial E|^{-1/2} \phi_0$.

$$k(E) = \frac{8\pi k}{3hc} N \left| \langle \phi_0 | D(R) | \phi_E \rangle \right|^2. \quad (4)$$

To describe the lowest rovibrational level, it is sufficient to approximate the ground electron state by a harmonic potential around the minimum R_0 ,

$V_g(R) = V_g + \frac{1}{2} V_g'(R - R_0)^2$, where $V_g \equiv V_g(R_0)$ and $V_g' \equiv dV_g(R)/dR|_{R=R_0}$. The ground-state wave function is then simply a Gaussian with the maximum at R_0 . The main contribution to the spectrum comes from the transitions around this minimum, say, from the classically allowed region of the width $\Delta R = 2\sqrt{\hbar/\mu\omega}$ around R_0 , where $\omega = \sqrt{V_g'/\mu}$ and μ is the reduced mass.

Within the narrow region ΔR around R_0 all the excited-state potentials can be fairly well described by the linearized form

$$V_e(R) = V_e + V_e'(R - R_0), \quad V_e \equiv V_e(R_0), \quad V_e' \equiv (dV_e(R)/dR)|_{R=R_0}. \quad (5)$$

If the excited potential $V_e(R)$ is steep around R_0 , the required matrix element [see equation (4)] is well described by

$$\langle \phi_0 | D(R) | \phi(E) \rangle = \frac{1}{\sqrt{|E|}} D(R_0) \phi_0(R_0). \quad (6)$$

where R_0 is the turning point of the excited-state potential for a given energy E . Writing the ground-state wave function, the energy of the lowest vibrational level, and the transition energy as

$$\phi_0(R) = \frac{1}{\pi^{1/4} \sigma^{1/2}} e^{-\frac{(R-R_0)^2}{2\sigma^2}}, \quad (7)$$

$$E_g = V_g + \frac{1}{2} \hbar \omega, \quad (8)$$

where $\sigma = \sqrt{\hbar/\mu\omega} = \Delta R/2$, one gets the expression (4) for the absorption coefficient in the form

$$k(v) = \frac{8\pi k}{3c} N [D(R_0)]^2 \frac{1}{w\sqrt{w}} e^{-\frac{2}{w^2}(E_g - h\nu)^2}, \quad (9)$$

where $E_g = V_g - E_0 = \hbar\nu_g$ and $w = |\mathcal{V}'\sigma|$. Here we also assumed D to be a slowly varying function of R , replacing it with $D(R_0)$.

